

Instructions – DC/AC

The files come in three different extensions, compatible with notepad: .RES, .SEB and .SER. The file .RES contains data for resistivity measurements, both AC and DC. The .SEB files contain all measurement points for Seebeck coefficient. The .SER files contain calculated values of Seebeck coefficients.

The typical format of the .RES file looks as follows:

```
HEOX CoCrFeMnNi 32 Cub02
27.10.2018 12:43:28
3.88 1.43 8.30 'Sample Dimensions A x B x Length [mm]
98 'Number of data
Point Time [s] Temp1[°C] Temp2[°C] Voltage[+V] Current[+A] Voltage[-V] Current[-A] Resistance[ohm] Resistivity[ohm x m]
1 10 23.639 23.561 1.76838E-01 2.60897E-05 ~~~~~ 6.77809E+03 4.53103094E+00
1 17 23.639 23.564 3.53734E-01 3.13822E-05 ~~~~~ 1.12717E+04 7.53498684E+00
1 24 23.636 23.573 5.30544E-01 3.31256E-05 ~~~~~ 1.60161E+04 1.07064966E+01
1 30 23.638 23.560 7.07710E-01 3.28047E-05 ~~~~~ 2.15734E+04 1.44214414E+01
1 37 23.641 23.567 1.06138E+00 3.41099E-05 ~~~~~ 3.11164E+04 2.08007620E+01
1 47 23.610 23.548 5.00928E-01 1.48843E-06 -4.98762E-01 -1.43735E-06 3.41682E+05 2.28408521E+02
1 55 23.618 23.540 7.50926E-01 2.19291E-06 -7.48552E-01 -2.19149E-06 3.42003E+05 2.28622641E+02
1 64 23.609 23.541 1.00067E+00 2.96739E-06 -9.98357E-01 -2.94914E-06 3.37871E+05 2.25860954E+02
1 72 23.602 23.530 2.00021E+00 5.84077E-06 -1.99795E+00 -5.86077E-06 3.41678E+05 2.28405557E+02
1 81 23.598 23.538 2.99940E+00 8.82873E-06 -2.99718E+00 -8.77577E-06 3.40628E+05 2.27703330E+02
2 2418 41.016 40.015 1.76828E-01 2.90994E-05 ~~~~~ 6.07667E+03 4.06214488E+00
2 2425 41.020 40.030 3.53704E-01 2.73470E-05 ~~~~~ 1.29339E+04 8.64608288E+00
2 2433 41.016 40.020 5.30496E-01 3.03198E-05 ~~~~~ 1.74967E+04 1.16962148E+01
2 2441 41.024 40.032 7.07632E-01 2.95298E-05 ~~~~~ 2.39633E+04 1.60190514E+01
2 2449 41.012 40.023 1.06126E+00 2.78561E-05 ~~~~~ 3.80980E+04 2.54678360E+01
2 2460 41.015 40.031 5.00721E-01 2.73246E-06 -4.98652E-01 -2.70269E-06 1.83872E+05 1.22915373E+02
2 2470 41.017 40.024 7.50691E-01 4.08501E-06 -7.48383E-01 -4.02823E-06 1.84769E+05 1.23514581E+02
2 2482 41.008 40.028 1.00039E+00 5.37246E-06 -9.98134E-01 -5.36715E-06 1.86089E+05 1.24396993E+02
2 2491 41.010 40.042 1.99967E+00 1.07260E-05 -1.99749E+00 -1.07949E-05 1.85733E+05 1.24159365E+02
2 2501 41.003 40.033 2.99863E+00 1.61598E-05 -2.99646E+00 -1.61688E-05 1.85442E+05 1.23964922E+02
```

Fig. 1. An exemplary .RES file.

Sample's dimensions

1st column - temperature point, since we use 5 different voltages for both AC and DC, the total number of rows is 10 - 1-5 are AC, 6-10 - DC

2nd column - time of the measurement

3rd column - temperature on the electrode with heater

4th column - temperature of the reference electrode

5th column - voltage

6th column - current

7th column - voltage for the second run with opposite polarization (only DC)

8th column - current for the second run with opposite polarization (only DC)

9th column - resistance (total)

10th column - resistivity (normalized with respect to the sample's size)

The .RES files are necessary to calculate the conductivity of our samples, and to plot the Arrhenius relation. The conductivity can be calculated as:

$$\sigma = \frac{l}{AR} \quad (1.1)$$

Since in our case, we already have the value normalized with respect to the sample's dimensions, we can use the values of resistivity from 10th column directly:

$$\sigma = \frac{1}{\rho} \quad (1.2)$$

Then, we have to use the Arrhenius equation:

$$\sigma T = \sigma_0 \exp\left(-\frac{E_a}{RT}\right) \quad (1.3)$$

Taking the natural logarithm:

$$\ln \sigma T = \ln \sigma_0 - \frac{E_a}{RT} \quad (1.4)$$

Now, what we have to do, is to plot this relation in a $\ln \sigma T(1/T)$ coordinates. Then, we will see that the relation has a linear character, which may change with the temperature. Usually, we observe 2 or 3 separate stages, characterized by different slopes. For each stage, we should perform a linear regression, obtaining in the process the value of energy of activation, characteristic for a given stage.

In each .RES file a whole history of the measurement is saved. That means that the data can be from two different runs: from room temperature to 1000 °C, and back from 1000 °C to room temperature.

The typical format for the .SEB file looks as follows:

```
HEOX CoCrFeMnNi 32 Cub01
24.10.2018 11:02:10
980 'Number of data'
Point Time[s] Temp1[°C] Temp2[°C] DeltaT[°C] Voltage[μV] Grad.No
1 339 20.29 20.29 0.01 92.469 1
2 347 21.19 20.30 0.89 -36.698 1
3 356 21.80 20.34 1.46 -175.164 1
4 366 22.56 20.39 2.17 -157.582 1
5 375 23.00 20.42 2.58 -253.893 1
6 383 23.56 20.45 3.11 -235.714 1
7 391 24.17 20.49 3.68 -468.111 1
8 400 24.51 20.54 3.97 -346.661 1
9 410 25.17 20.59 4.58 -433.935 1
10 419 25.44 20.66 4.78 -507.536 1
11 3521 40.69 39.76 0.93 -168.489 2
12 3530 41.42 39.80 1.62 -207.078 2
```

Fig. 2. An exemplary .SEB file.

1st column - measurement point (for each temperature point, 10 measurements are performed)

2nd column - time of the measurement

3rd column - temperature on the electrode with heater

4th column - temperature of the reference electrode

5th column - the difference between these temperatures ΔT

6th column - measured potential ΔV

7th column - number of temperature point

The Seebeck coefficient S is defined as:

$$S = - \frac{\Delta V}{\Delta T} \quad (1.5)$$

To calculate its value for a given temperature point, we have to plot the ΔV in a function of ΔT and calculate the slope of the resulting function:

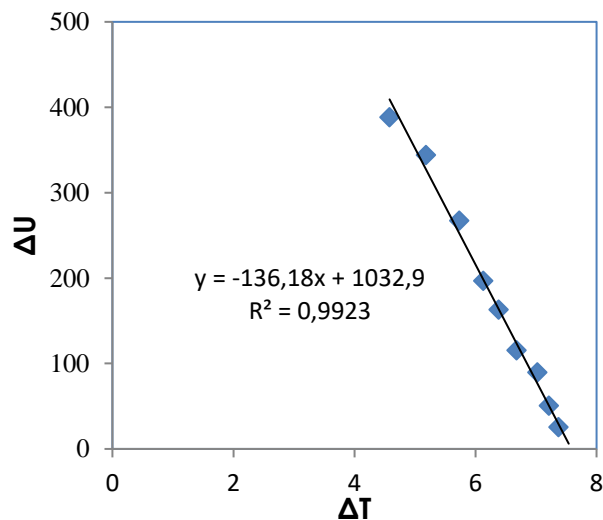


Fig. 3. An exemplary determination of Seebeck coefficient.

Similarly as previously, in each .SEB file a whole history of the measurement is saved. That means that the data is in fact for two different runs: from room temperature to 1000 °C, and back from 1000 °C to room temperature.

The results from procedure shown in the Fig. 3 are compiled for all temperature points in .SER file:

Typical format for the .SER file looks as follows:

```
HEOX CoCrFeMnNi 32 Cub01
24.10.2018 11:02:10
98      'Number of data
Point  Temp1[K]  Seebeck Coef.[μV/K]  Linear Correlation
1      20.29    -1.15788982647E+0002  -0.9604344083
2      39.76    -1.19591305869E+0002  -0.9960943037
3      58.50    -1.27278201104E+0002  -0.9986102194
4      77.43    -1.23725613288E+0002  -0.9939513105
5      96.75    -1.18953400778E+0002  -0.9977514413
6      116.78   -1.57122387225E+0002  -0.7787262770
7      136.48   -1.11472774304E+0002  -0.9956008150
8      155.98   -1.13486457611E+0002  -0.9950321658
9      176.40   -1.13519667196E+0002  -0.9945490583
10     196.98   -1.14162716573E+0002  -0.9951667482
11     217.56   -1.12525713046E+0002  -0.9969780447
```

Fig. 4. Exemplary .SER file.

1st column - temperature point

2nd column - temperature

3rd column - value of Seebeck coefficient

4th column - value of the R^2 correlation factor

A typical way of determining the potential of a given material in the thermoelectric application, is to calculate its figure of merit ZT :

$$ZT = \frac{\sigma S^2 T}{\kappa} \quad (1.6)$$

where: σ - electrical conductivity, S - Seebeck coefficient, T - temperature, and κ - thermal conductivity (remember about the common units!).

Except for the last property, all other can be extracted from the .RES and .SEB file. The κ can be calculated for the spinel samples 32 and 36, using the following equation:

$$\kappa [Wm^{-1}K^{-1}] = 443174T^{-1.8692} + 7.3970 \quad (1.7)$$

What is to be done:

- calculate conductivity and plot the Arrhenius relation for 4 different measurements runs (remember, we have 2 per file)
- calculate the energies of activation for each plot and for each distinct stage (if two or more different slopes are present)
- plot the temperature dependence of Seebeck coefficient for each of the measurements (remember, that the .RES files and .SER files of the same name refer to exactly the same measurement)

- using the values of σ , S , and T , as well as relations (1.6) and (1.7), calculate the ZT for samples 32 and 36.

In the report there should be:

- 4 figures with Arrhenius plots
- 4 tables with the values of energies of activation for each measurement
- 4 tables and 4 figures with $ZT(T)$ for each of the measurements - only samples 32 and 36.
- a few words of comment, maybe comparison with some literature values - "*impress me*"

32 - $(\text{Co,Cr,Fe,Mn,Ni})_3\text{O}_4$

36 - $(\text{Co,Cr,Fe,Mn,Mg})_3\text{O}_4$

Instructions – EIS

In the excel file “117” there are 4 sheets, each with 3 columns: frequency (f), resistivity (Z') and reactance (Z''). The results were measured at four consecutive temperature points (150, 200, 250 and 300°C, as indicated in the sheets' names) for the $(\text{Ce,Gd,Nd,Pr,Sm})\text{O}_{2-\delta}$ material. In the next file “EIS”, a solver with an implemented solution for the following equivalent circuit is prepared:

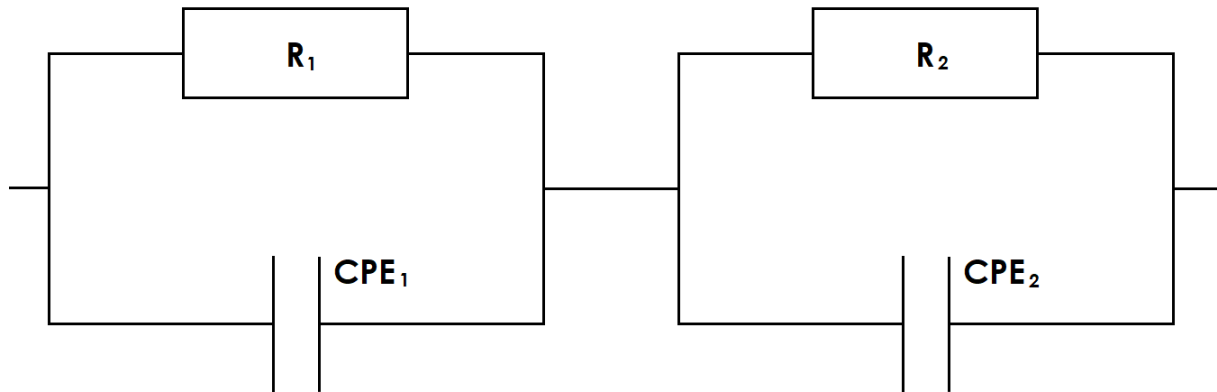


Fig. 4. Equivalent circuit for the $(\text{Ce,Gd,Nd,Pr,Sm})\text{O}_{2-\delta}$.

As can be seen, we can distinguish two R-CPE circuits. The first of them describes the properties of the bulk, while the second of the grain boundaries in the material. The CPE abbreviation stands for the constant phase element, which describes non-ideal behavior of the capacitor:

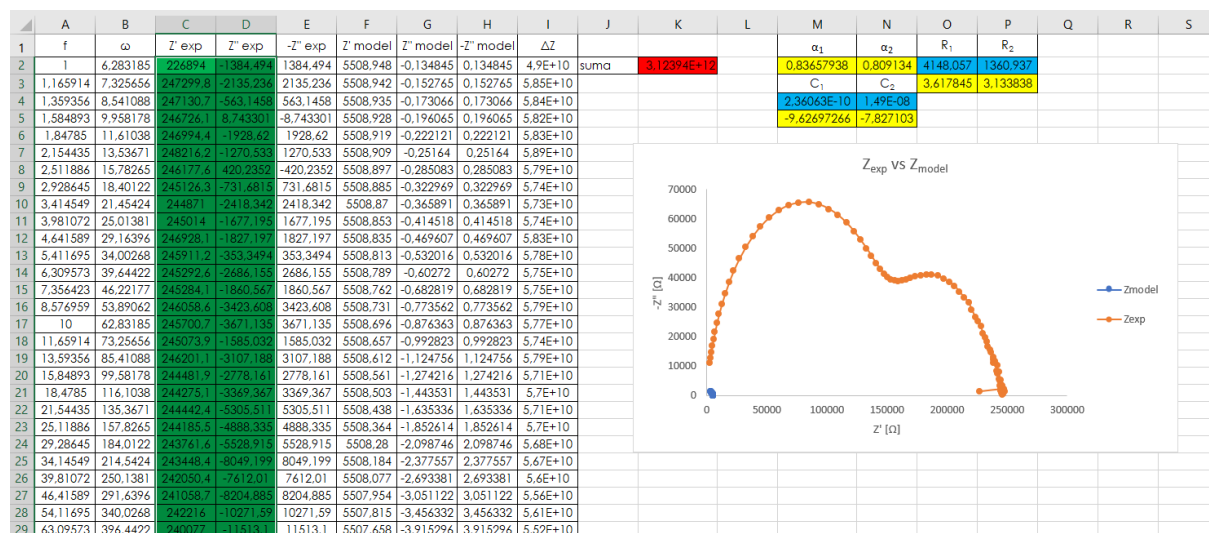
$$Z_{CPE} = \frac{1}{C(j\omega)^\alpha} \quad (1.8)$$

Where C – non-ideal capacity, α – constant. For the values of α equal to 1, our CPE behaves like an ideal capacitor, for the values equal to 0 – like a resistor.

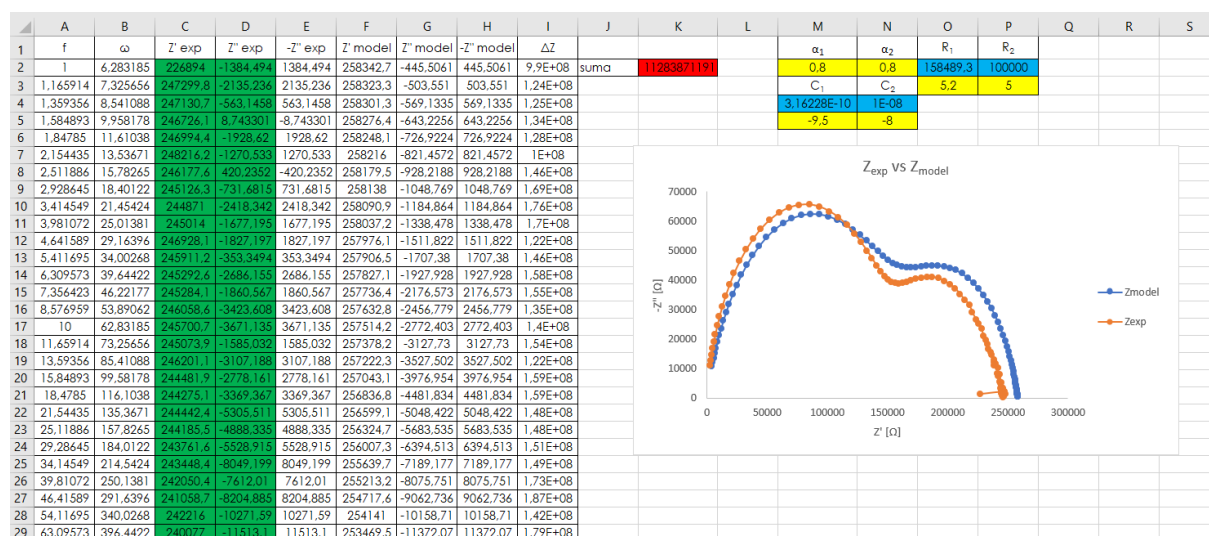
In order to find the values of resistivity and capacity characteristic for the bulk and grain boundary conductivity, we have to run an optimization of our experimental results. To perform this operation,

we have to copy the Z' and Z'' values from the “117” to columns C and D in “EIS” file (highlighted in green). Then, we can start the comparison of the theoretical values with the experimental ones. In columns M to P, are the starting parameters for our optimization (highlighted in yellow). All values, except for α , are given as exponents, based on which are calculated the final values (blue).

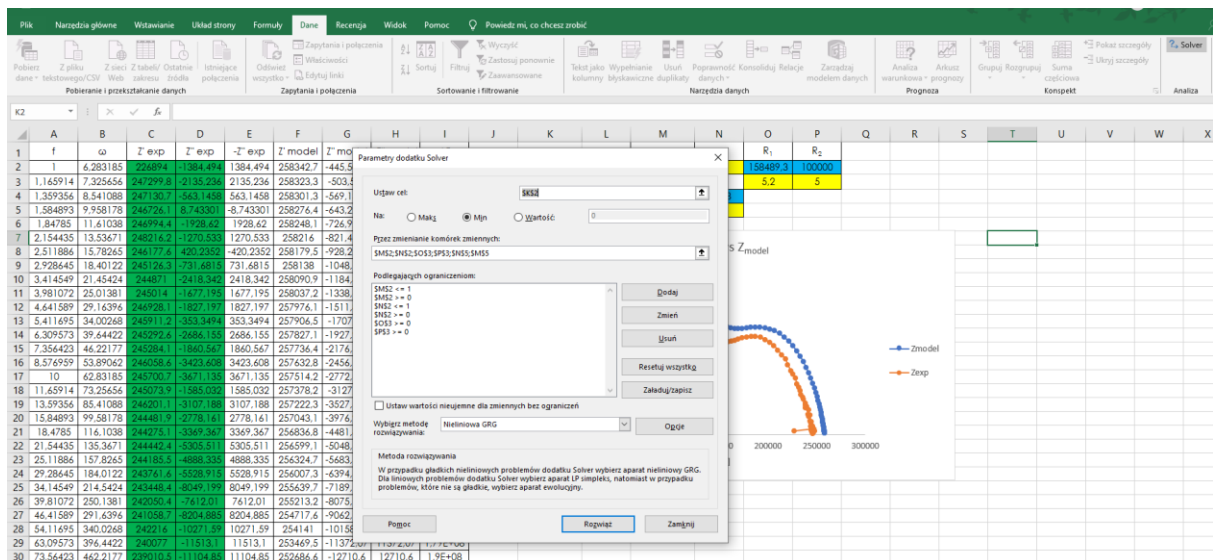
So, first, we copy the values from “117” to green columns:



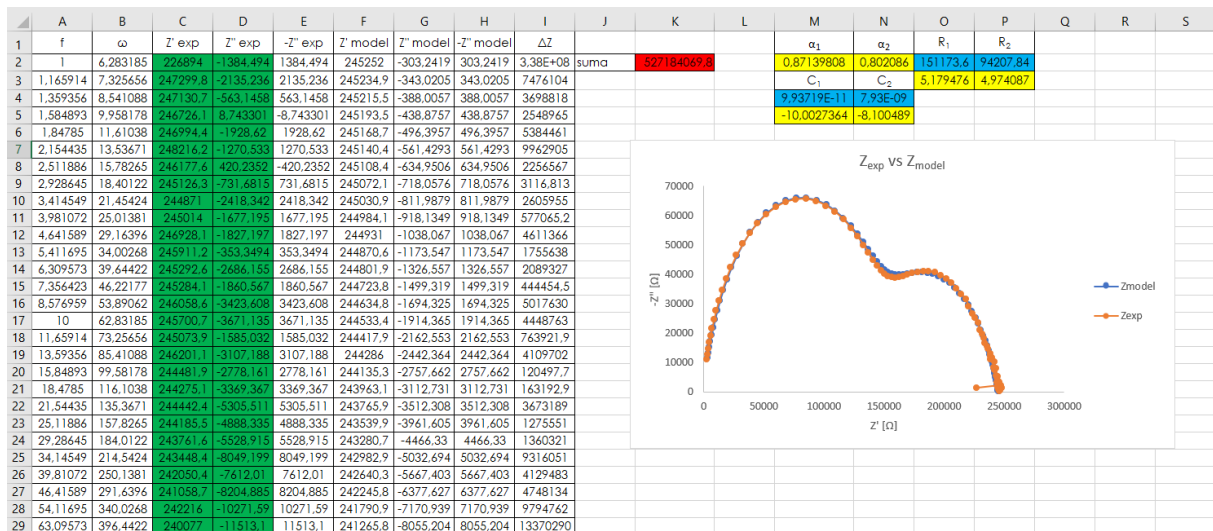
Then, we try to set the values in yellow cells, to get our “Zmodel” plot to similar ranges as the experimental “Zexp” (just similar, not identical):



Then, we use the “solver” plug-in (“dane” menu in excel):



All the setups are visible in the above figures. Our optimization is based on the least squares element method, implemented in the column I and summarized in our target function in cell K2 (red). After solving the task, we should get the optimized plots and final values (blue):



The value in cell O3 represents the resistivity of the bulk, while the value in cell P3 describes the resistivity of the grain boundaries. Using the eq. (1.1), we can then calculate the conductivity for both cases, and plot the Arrhenius plots according to eqs. (1.3) and (1.4). The dimensions for the sample are as follows:

- $l = 0.97 \text{ mm}$
- $A = 52.17 \text{ mm}^2$

What is to be done:

- Perform the optimization for all 4 temperatures, save the optimized files for each of them and attach them to the report.
- Gather all plots and optimized values and put them into the report.
- Recalculate the obtained resistivity values into conductivity and create Arrhenius plots, one for the bulk and one for the grain boundaries.
- Calculate the energies of activation for both mentioned processes.