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Groundwater monitoring

title: Hydrogeochemical monitoring in a coastal aquifer subject to an intense seawater abstraction. The case of the River Andarax delta (Almería, SE Spain)

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INTRODUCTION

The delta of the River Andarax is situated on the coastal strip of the Detritic Aquifer of the Lower Andarax (Almería, SE Spain). The delta aquifer deposits consist of 100 m thickness of alternating sands, gravels and lutites. The desalination plant installed in this delta aquifer abstracts a large volume of seawater from coastal boreholes. A monitoring network was designed close to the water collection area, consisting of three piezometer clusters (PI, PII and PIII), 500 m apart, each including four simple piezometers: one that is slotted over its entire permeable length, and the remaining three with a 1-2 m slotted section at particular depths in the zones of fresh water, salt water and mixing. The piezometers in each cluster are positioned at different depth, depending on the position of the fresh and salt water bands in each monitoring group (Jorreto el al., 2009).

The fully-slotted piezometers were sampled in order to characterize the hydrochemistry of the area. Samples were taken at different depths corresponding to the fresh water band (12–15 m depth), band of transition (25–30 m deep) and saline band (38–55 m deep). The remaining piezometers were sampled over their slotted length. Overall, 22 samples were taken.

RESULTS AND DISCUSSION

The waters sampled from the piezometer network exhibit varying salinity (6.8 to 51.8 mS/cm) and a chloride facies, typical of a coastal aquifer with marine intrusion. Such wide variability in the different piezometers allows the hydrogeochemical zoning of the aquifer (fresh water, mixing zone seawater) to be determined. However, this zoning was not recorded in the fully slotted piezometer (PII1), sampled at five different depths. This piezometer gave relatively low ion contents compared to the others, except for nitrate, which was higher. Samples from this piezometer also showed a narrower dispersion than the other piezometer PII1 implies a salinity similar to that taken from the five sampling depths in piezometer Water was less saline (9.7 mS/cm). This homogeneity is interpreted as being a consequence of the abstraction of seawater in the boreholes closest to PII1, which must be affecting the situation of the interface in this piezometer. As seawater is abstracted, the interface descends – this phenomenon is detected from the temperature and electrical conductivity of water in the fully slotted piezometers. (Jorreto et al., 2006; Jorreto et al., 2009). As a result, the salinity of the water column in the borehole becomes close to that found in the upper aquifer levels.

	EC	Са	Mg	Na	К	Cl	HCO ₃	SO ₄	NO 3	Br	Sr
Standard Deviation	22.3	58	539	6285	188	9759	160	802	20	27	2
Mean	37.1	511	956	9437	255	14669	365	2591	23	41	8
Standard Deviation	4.2	21	83	853	21	1564	3	97	12	4	0
Mean	13.1	430	352	2399	49	4130	344	1602	83	11	5
Standard Deviation	20.8	235	568	5794	146	10389	36	867	53	29	4
Mean	32.2	786	946	7956	174	14438	337	2690	41	40	11
	Standard Deviation Mean Standard Deviation Mean Standard Deviation Mean	ECStandard Deviation22.3Mean37.1Standard Deviation4.2Mean13.1Standard Deviation20.8Mean32.2	ECCaStandard Deviation22.358Mean37.1511Standard Deviation4.221Mean13.1430Standard Deviation20.8235Mean32.2786	EC Ca Mg Standard Deviation 22.3 58 539 Mean 37.1 511 956 Standard Deviation 4.2 21 83 Mean 13.1 430 352 Standard Deviation 20.8 235 568 Mean 32.2 786 946	EC Ca Mg Na Standard Deviation 22.3 58 539 6285 Mean 37.1 511 956 9437 Standard Deviation 4.2 21 83 853 Mean 13.1 430 352 2399 Standard Deviation 20.8 235 568 5794 Mean 32.2 786 946 7956	EC Ca Mg Na K Standard Deviation 22.3 58 539 6285 188 Mean 37.1 511 956 9437 255 Standard Deviation 4.2 21 83 853 21 Mean 13.1 430 352 2399 49 Standard Deviation 20.8 235 568 5794 146 Mean 32.2 786 946 7956 174	EC Ca Mg Na K Cl Standard Deviation 22.3 58 539 6285 188 9759 Mean 37.1 511 956 9437 255 14669 Standard Deviation 4.2 21 83 853 21 1564 Mean 13.1 430 352 2399 49 4130 Standard Deviation 20.8 235 568 5794 146 10389 Mean 32.2 786 946 7956 174 14438	EC Ca Mg Na K Cl HCO3 Standard Deviation 22.3 58 539 6285 188 9759 160 Mean 37.1 511 956 9437 255 14669 365 Standard Deviation 4.2 21 83 853 21 1564 3 Mean 13.1 430 352 2399 49 4130 344 Standard Deviation 20.8 235 568 5794 146 10389 36 Mean 32.2 786 946 7956 174 14438 337	EC Ca Mg Na K Cl HCO3 SO4 Standard Deviation 22.3 58 539 6285 188 9759 160 802 Mean 37.1 511 956 9437 255 14669 365 2591 Standard Deviation 4.2 21 83 853 21 1564 3 97 Mean 13.1 430 352 2399 49 4130 344 1602 Mean 20.8 235 568 5794 146 10389 36 867 Mean 32.2 786 946 7956 174 14438 337 2690	EC Ca Mg Na K Cl HCO3 SO4 NO3 Standard Deviation 22.3 58 539 6285 188 9759 160 802 20 Mean 37.1 511 956 9437 255 14669 365 2591 23 Standard Deviation 4.2 21 83 853 21 1564 35 977 12 Mean 13.1 430 352 2399 49 4130 344 1602 83 Standard Deviation 20.8 235 568 5794 146 10389 366 867 53 Mean 32.2 786 946 7956 174 14438 337 2690 41	ECCaMgNaKClHC03SO4NO3BrStandard Deviation22.358539628518897591608022027Mean37.151195694372551466936525912341Standard Deviation4.22183853211564397124Mean13.1430352239949413034416028311Standard Deviation20.8235568579414610389368675329Mean32.278694679561741443833726904140

Table 1. Means and standard deviations of ion content (mg/L) and electrical conductivity (mS/cm) in the fully slotted piezometers.

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